

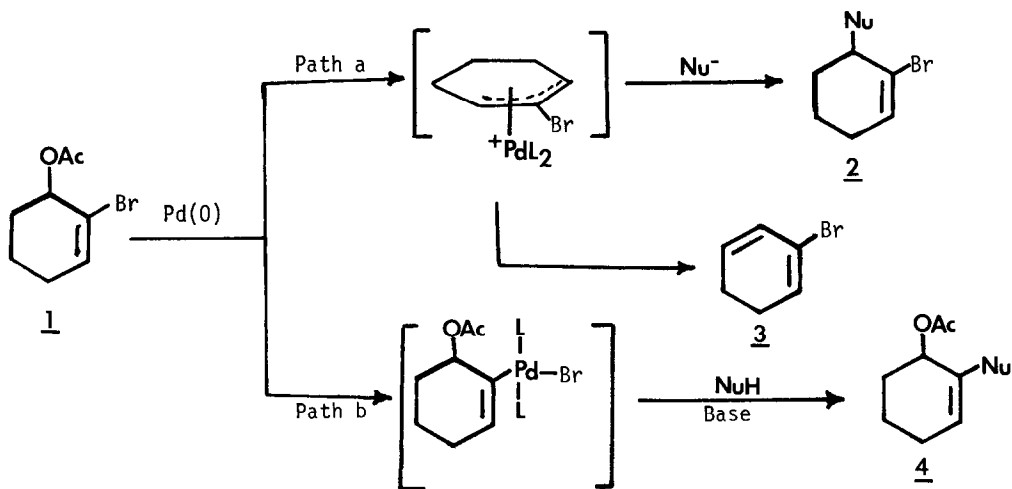
A NEW TYPE OF FUNCTIONAL GROUP DIFFERENTIATION IN PALLADIUM-CATALYZED REACTIONS

Godson C. Nwokogu

Department of Chemistry, University of Maine, Orono, Maine 04469, U.S.A.

Abstract: Products of palladium-catalyzed ethynylations of various 2-bromoallyl acetates show that the formation of σ -vinylpalladium intermediates, instead of π -allylpalladium complexes, is preferred in such substrates.

Palladium-catalyzed nucleophilic alkylations of simple allyl acetates by way of π -allylpalladium complexes¹ and the coupling of aryl or vinyl halides with nucleophiles through σ -vinylpalladium intermediates² are well-known reactions. For 2-haloallyl acetates (e.g. 1, Scheme 1) which bear both the allyl acetate and vinyl bromide moieties such that they affect each other, palladium-catalyzed reaction could lead to any or all of the product types 2, 3 or 4 through the two probable reaction pathways a and b. We did not find any previous report concerning this type of chemoselectivity in palladium-catalyzed reactions of substrates like 1.



Scheme 1

We report here that various 2-bromoallyl acetates, under palladium-catalyzed reaction conditions, reacted with acetylenes according to path b only. No product types expected from

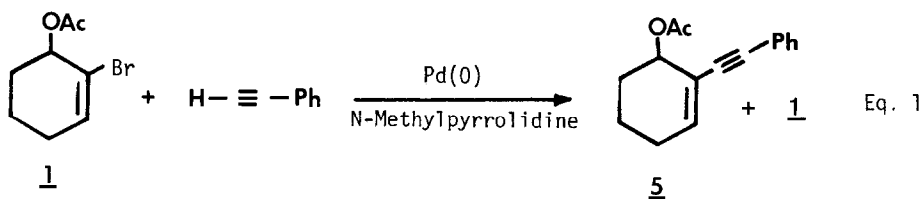
a π -allylpalladium intermediate, like 2 or 3 for substrate 1 in Scheme 1, were observed.

Refluxing a solution of 10 mmol of 2-bromo-3-acetoxycyclohexene 1,³ 2 mol % of various palladium catalyst systems (Table 1) and 15 mmol of phenylacetylene in 10 mL of N-methylpyrrolidine for 6-8 h gave 24-42% of the acetylene-coupled product 5 and some

Table 1. Palladium-catalyzed Ethynylation of 2-Bromo-3-acetoxycyclohexene 1.

Catalyst () ^a	Yield % ^b	
	<u>5</u>	Recovered <u>1</u>
Pd(OAc) ₂ /PPh ₃ (1:2)	24	47
(PPh ₃) ₂ PdCl ₂ /CuI (2:1)	25	45
(PPh ₃) ₄ Pd	42	27

(a) Mol ratio (b) Of isolated compounds, not optimized.



unreacted bromoacetate 1 only (Eq. 1).

Other examples of this chemoselectivity are given in Table 2. Tetrakis(triphenylphosphine)palladium(0) was used for all reactions on the table except for Entry 2 for which (PPh₃)₂PdCl₂/CuI gave a better result.

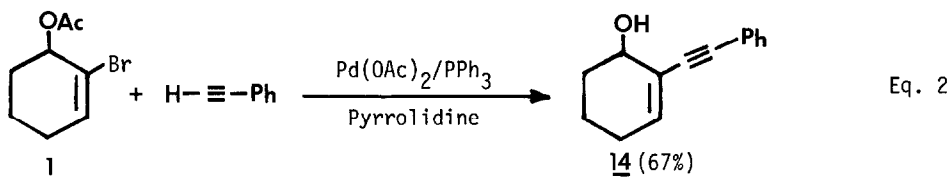
Verification of the absence of a π -allylpalladium intermediate in these reactions was based on the fact that in the absence of a nucleophile, a π -allylpalladium complex should undergo elimination to give a diene.⁷ With 1 under this reaction condition, elimination should afford 2-bromocyclohexadiene 3 (Scheme 1). When 1 was refluxed in N-methylpyrrolidine for 6 h with 2 mol % of (PPh₃)₂PdCl₂ and 1 mol % of CuI only, it was recovered unreacted in 91%. Recovery of unreacted substrate was also high in reactions with unreactive acetylenes (see Entry 4, Table 2). These very high recoveries of 2-bromoallyl acetates under conditions that favor elimination in simple allyl acetates are consistent with the absence or insignificance of the π -complex reaction route in these reactions.

We obtained further evidence of the unusual reactivity of 2-haloallyl acetates from their palladium-catalyzed reactions in pyrrolidine. For example, refluxing 1 with phenylacetylene in pyrrolidine using 2 mol % of Pd(OAc)₂ and 4 mol % of PPh₃ for 2-4 h gave the enynol 14 in 67% (Eq. 2). Repeating the reaction without the acetylene resulted in the isolation of the

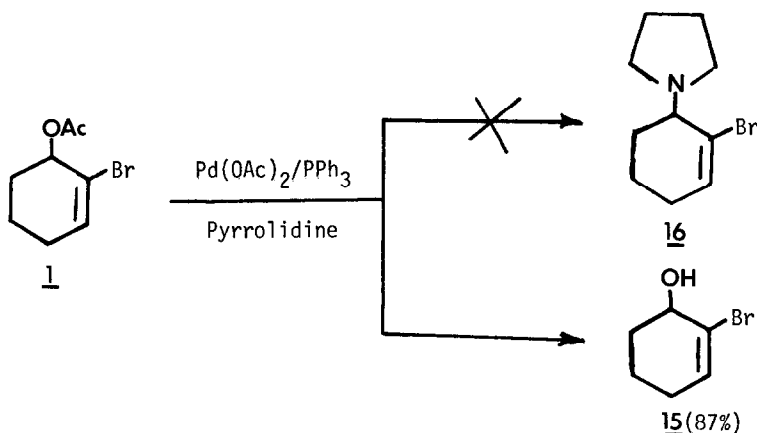
Table 2. Palladium-catalyzed Ethynylation of 2-Bromoallyl Acetates.

Entry	Substrate	Acetylene	Reaction Time(h)	Product ⁴ (%) ^a	Recovered Substrate (%) ^a
1 ^b		$\text{H}-\equiv-\text{Si}\equiv$	3		15
2	<u>1</u>	$\text{H}-\equiv-\text{C}(\text{OH})\text{CH}_3$	4		27
3		$\text{H}-\equiv-\text{Ph}$	0.7		0
4	<u>8</u>	$\text{H}-\equiv-\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$	12	--	93
5		$\text{H}-\equiv-\text{Ph}$	1.5		48
6 ^b		$\text{H}-\equiv-\text{Si}\equiv$	15		22

(a) Isolated compounds, yields not optimized (b) Reaction in a sealed tube at 80°C.



bromoalcohol 15⁸ in 87% yield. The amination product 16, which would be expected from a π -



allylpalladium intermediate,⁹ was not observed.

The above results represent the first examples of this type of chemoselectivity in palladium catalyzed reactions. They also implicate an unusual effect of a 2-halosubstituent on the ease of fission of the allylic carbon-oxygen bond in allyl acetates.

Acknowledgements: We wish to thank Professor Harold Hart for helpful suggestions and Professor K. Biemann for high resolution mass spectral analyses.

References and Notes

1. Trost, B.M., *Tetrahedron*, **32**, 2615 (1977).
2. Heck, F.R., Dieck, H.A., *J. Organomet. Chem.*, **93**, 259 (1975).
3. Sandler, S.R., *J. Org. Chem.*, **32**, 3876 (1967).
4. All new compounds had correct chemical analyses (elemental or high resolution mass spectral). For 11: mp 146-7°C; ¹H NMR(CCl₄): δ 7.20(m, 15H), 4.80(s, 2H), 2.00(s, 3H); Calc'd for C₂₅H₂₀O₂: C, 85.20; H, 5.72. Found: C, 84.92; H, 5.64.
5. 2-(Phenylethynyl)cyclohepta-1,3-diene, a secondary product from 9 was also obtained in 16%. Its quantity depended on reaction time. For example, after 3 h, it was isolated in 60% while 9 was obtained in 40% yield.
6. Prepared in 52% yield overall from norbornylene by refluxing its dibromocarbene adduct, prepared according to: Moore, W.R., Moser, W.R., LaPrade, J.E., *J. Org. Chem.*, **28**, 2200 (1963), in glacial AcOH/NaOAc.
7. Trost, B.M., Verhoeven, T.R., Fortunak, J.M., *Tetrahedron Lett.*, **25**, 2301 (1979).
8. Sonnenberg, J., Winstein, S., *J. Org. Chem.*, **27**, 748 (1962).
9. Atkins, K.E., Walker, W.E., Manyik, R.M., *Tetrahedron Lett.*, **45**, 382 (1970).

(Received in USA 25 April 1984)